

# Introductory Chemistry

## Chem. 10.61 Syllabus

Lecture: MW 5:30 PM – 7:20 PM -- Room SC2210

Lab: M 7:30-10:20 PM -- Room SC2210

**{Be Kind}**

**Instructor:** Dr. James Maxwell, email: maxwelljames@fhda.edu (Best Way to contact me!)  
Mobile phone: (773) 454-7779 (texts also),

**Office Hours:** M & W: 4-5pm, Second Floor SC1 or by appointment.

**Description:** An Introduction to the discipline of Chemistry, including chemical laboratory techniques and methods and a survey of important chemical principles. This course emphasizes chemistry as a subject of scientific inquiry and is designed to give the student a general appreciation for chemistry as a science

**Prerequisites:** Advisory: English Writing 211 and Reading 211 (or Language Arts 211), or English as a Second Language 272 and 273. Mathematics 212 or the equivalent.

**Evaluation:** Your grade will be based on your performance in the following:

10 best Quizzes (10 pts. each)		100 points
8 Lab reports (20 pts. each)	160	
Lab Clean-up	20	
1 Lab Final Exam (100 pts.)	100	
3 Exams (100 pts. each)	300	
<u>1 Final (200 pts.)</u>	<u>200</u>	
<b>Total</b>	<b>880 points</b>	

Letter grades will be assigned according to the *approximate* scale:

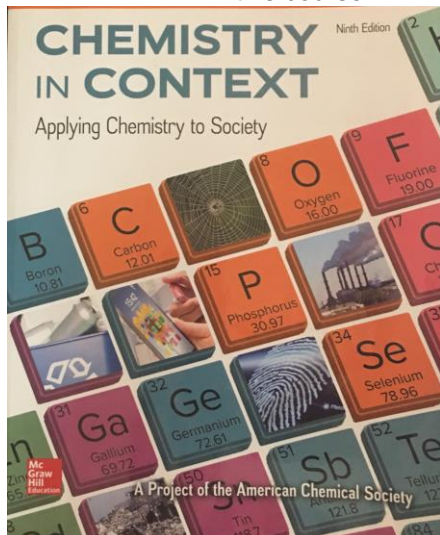
A	90%
B	80%
C	70%
D	50%
F	< 50%

**Attendance:** You MUST attend the first class and lab or you will be dropped from the course. Your attendance is urged for all lectures and required for all quizzes, exams and labs. Unexcused exam, quiz and lab absences score a zero (0) points. It is the responsibility of the student to contact the professor regarding missed work. If an absence is anticipated, the student should make arrangements to complete the missed assignments prior to the absence. In an emergency, it is the student's responsibility to contact the instructor within one class period of an exam. *There are no laboratory make-up days.* **Please sign the attendance sheet each class and lab.**

**Quizzes:** Quizzes will be given during class as scheduled in syllabus, and will have a time limit. Answer keys will be available after the quiz. *If you miss the quiz, you will **not** have a chance to make it up.* The best 10 quiz scores will be used in determining your final grade. The three lowest quizzes will be dropped except for Quiz 12 (20 points). That quiz cannot be dropped.

**Exams:** There will be three exams and one final exam. You must bring your own scientific calculator, pencil or pen and eraser for exams. Cell phones may not be used at any time during the exam. Scientific Calculators may be used. **No Cell Phones during Exam! Answer Keys will be available after the exam.**

**Text:** **Chemistry for Changing Times:** *Applying Chemistry to Society*, 9e, by Fahlman, Roberts, Kirk, Bentley, Daubenmire, Ellis, and Mury. ISBN: 978-1-259-692015-1. You may be able to rent your text at [www.vitalsource.com](http://www.vitalsource.com). Unfortunately, any previous edition of this text is distinctly different than this edition. Purchasing any previous edition will be problematic with regards to following the contents of this course.

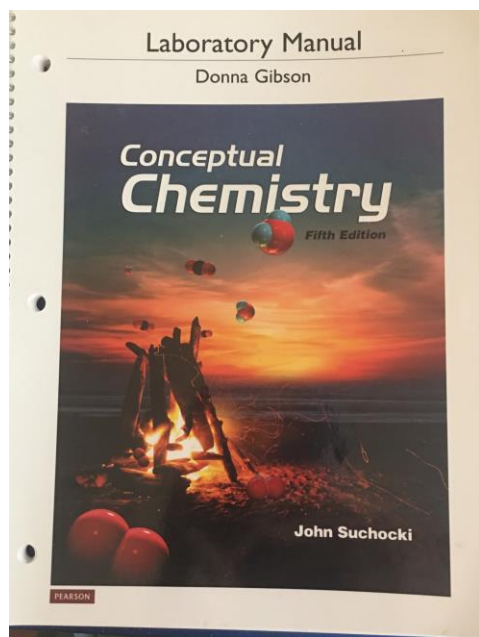


**Schoology:** We will use Schoology.com to communicate course materials, Keys to Quizzes and Exams and other important information. Please go to [www.Schoology.com](http://www.Schoology.com) and use code **56KM-8RF8-8NVN2** to register for Chem 10.61 Fall 2019

**Lab Safety Statement:** Please sign and return this **ASAP**. The instructor will give you a copy.

**Supplemental Test:** *Chemistry-Atoms First*. This **free** text is available in the Course Studio. We will use a few chapters, as we need them to help with lecture and lab topics.

**Lab Text:** **Laboratory Manual for Conceptual Chemistry**, 5e, 2014, Prentice Hall, by Donna Gibson & John Suchocki



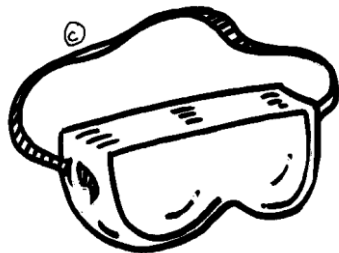
**Lab Report:** Your notebook will be graded during lab and returned to you before you leave lab. In **NOTEBOOK 1** write Labs 1,3,5,7 and in **NOTEBOOK 2** write Labs 2,4,6. Notebooks to be sewn binding, available at office supply, grocery stores or drug stores, about \$3 each (shown here). Lined pages please. Choose a color that is **NOT** black. See images below. **Labs will be due for grading the week following the completion of the experiment. I will grade them during lab and return them to you.**



1. Name and Contact Information, include an email or phone number
2. Number each page from front to back in upper right hand corner. Number each side of the page.
3. Table of contents on First Page: Experiment name and Page Number. You will fill this in as the Quarter proceeds. I recommend recording Labs 1,3,5,7, and 9 in Book I and Labs 2,4,6, and 8 in Book II. I will grade one book as you work in the other.
4. Experiment Name
5. Experiment Objective (can summarize)
6. Materials list
7. Brief Discussion or Theory
8. Procedure: Summarize, but must be able to follow in class to perform the experiment.
9. Data Table as it appears in the lab book.
10. Discussion of Results including possible sources of error
11. Questions and Answers: Must be in Full Sentences
12. After each page is completed, sign and date the bottom of each page.



**Eye Protection:** You **MUST** wear full goggles and not safety glasses. Without them, you may not participate in lab and will receive a grade of zero for that lab. See illustration below. **Only the Safety Goggles sold at the DeAnza bookstore are allowed. No substitutions.**

**Important Dates:**

September 23	First day of fall quarter
October 5	Last day to <a href="#">add classes</a>
October 6	Last day to <a href="#">drop classes</a> for full refund or credit
October 6	Last day to <a href="#">drop classes</a> without a W
October 18	Last day to request " <a href="#">Pass/No Pass</a> " for 12-week classes
November 11	<b>Veterans Day Holiday:</b> Campus closed
November 15	Last day to <a href="#">drop classes</a> with "W"
Nov 28-Dec 1	<b>Thanksgiving Holiday:</b> Campus closed
December 1	Last day to file for fall <a href="#">degree or certificate</a>
December 9-13	<b>Final exams:</b> Our Final Exam: Monday 9 December at 6;15-8:15 pm
December 13	Last day of fall quarter

**Changes to Syllabus:** **This syllabus may/will change according to the instructor and the needs of the class.**

Changes will be announced and published in the Course Studio, but you must read and follow the changes. Please check with the syllabus posted (latest update will be posted and dated) or with your instructor.

**Lab Housekeeping Rules: A clean lab is a safe lab!**

1. **Wear protective eyewear at all times during experimentation.** Your instructor will let you know if you may remove your protective eyewear.
2. Clean up all broken glass immediately with broom and dust pan and dispose in special container
3. Clean all chemical spills immediately. Notify your instructor who will advise or help you. Dispose all chemicals in the designated container.
4. Do not pour any chemicals down the drain. Dispose only in designated containers as indicated by your instructor.
5. Go to the stock chemical container. DO NOT take the container to your desk. DO NOT pour used chemicals back in stock container. Only take what you need to minimize waste. Un-needed chemicals must be taken to disposal container.
6. Always carefully read the label to insure you will use the correct chemical reagent.
7. Do not wear shorts, open-toed shoes or diaphanous clothing on lab days.
8. Clean your glassware and your lab bench before you leave the lab. The lab should always be in better shape than when you found it.
9. It Is Your Job/Responsibility. It doesn't matter if your did not make the mess, your should clean it up.
10. The balance room should be left in immaculate order: Clean, All balances in perfect order, all chairs under the desks. NO EXCEPTIONS!
11. Wash your hands before you leave lab, no matter what experiment was performed.
12. The lab is always under observation and any unkempt areas will be recorded and reported to me and the head of the department and then to you.
13. Thanks for keeping these concepts and any other concepts necessary for a safe and clean lab.

**How to succeed in Chemistry 10:**

1. Always come to class prepared. Please read your assignments and work as many homework problems as you can.
2. Form a study group with your classmates. Work homework together.



- Attend class and pay attention. Ask questions.
- Stay current with assignments.
- Stay current with lab: have pre-lab complete before class. Turn in lab reports on time.
- Get help before it is too late. Use your instructor's office hours or get help from the tutoring center on campus.
- Keep your life in balance: Study, Play, Relax.

## CHANGE YOUR WORDS CHANGE YOUR MINDSET

**FIXED**

I'M JUST NOT GOOD AT THIS!


I GIVE UP!

I'll never be as smart!

THIS IS TOO HARD!

I'm never going to get this!

I CAN'T MAKE THIS ANY BETTER!



**GROWTH**

I'LL USE SOME OF THE STRATEGIES I'VE LEARNED!

I CAN ALWAYS IMPROVE. I'LL KEEP TRYING!

This may take some time and effort!

I NEED TO FIGURE OUT WHAT I DID WRONG AND GET SOME HELP!

I'M ON THE RIGHT TRACK!

WHAT AM I MISSING?

Date Mon	Lecture + Laboratory	Lecture
23 Sept	Intro to Course and Lab Lecture: Chap 1: Portable Electronics: The Periodic Table in the Palm of Your Hands; <b>Lab: Check-in</b>	Lecture: Ch 1 cont. Ch 2: The Air We Breathe
30 Sept	Lecture: Ch 2 cont. Ch 3: Radiation from the Sun <b>Quiz 1: Ch 1</b> <b>Lab 1: Taking Measurements</b> <b>Lab Safety Statement DUE</b>	Lecture: Ch 3 cont. Ch 4: Climate Change <b>Quiz 2: Ch 2</b>
7 Oct	Lecture: Ch 4 cont. <b>Quiz 3: Ch 3</b> <b>Lab 2: Percent Water in Popcorn</b>	Lecture: Review Exam 1 <b>Quiz 4: Ch 4</b>
14 Oct	Lecture: <b>Exam 1;/Chaps 1-4</b> <b>Lab 3: Electron Dot Structures</b>	Lecture: Ch 5: Energy from Combustion
21 Oct	Lecture: Ch 5 cont.; Ch 6: Energy from Alternative Sources <b>Lab 4: Molecular Shapes</b>	Lecture: Ch 6 cont. Ch 7: Energy Storage <b>Quiz 5: Ch 5</b>
28 Oct	Lecture: Ch 7 cont. Ch 8: Water Everywhere: A Most Precious Resource <b>Quiz 6: Ch 6</b> <b>Lab 5: Solutions</b>	Lecture: Ch 8 cont. Review for Exam 2 <b>Quiz 7: Ch 7</b> <b>Quiz 8: Ch 8</b>
4 Nov	Lecture: <b>Exam 2: Chaps 5-8</b> <b>Lab 6: Upset Stomach</b>	Lecture: Ch 9: The World of Polymers and Plastics
11 Nov	<b>Holiday: Veterans Day</b>	Lecture: Ch 9 cont.; Ch 10: Brewing and Chewing

		<b>Quiz 8: Ch 8</b>
18 Nov	Lecture: Ch 10 cont. Ch 11: Nutrition <b>Quiz 9: Ch 9</b> <b>Lab 7: Organic Molecules</b>	Lecture: Ch 11 cont. Ch 12: Health and Medicine <b>Quiz 10: Ch 10</b>
25 Nov	Lecture: Ch 12 cont. Ch 13: Genes and Life <b>Quiz 11: Ch 11</b> <b>Quiz 12: Ch 12</b> <b>Lab 8: DNA Capture from a Banana</b>	Lecture: Ch 13: Cont. Review for Exam 3 and Lab Final <b>Quiz 13: Ch 13</b>
2 Dec	Lecture: Final Review Chaps. 1-13 <b>Quiz 14: Mini-Final (Take Home- Cannot be dropped, 20 pts)</b> <b>Lab: Lab Final</b> <b>Lab Check-Out</b>	<b>EXAM 3: Chaps 9-13</b> <b>Quiz 14: DUE</b>
9 Dec	<b>Final Exam: Chaps 1-13</b> <b>6:15 pm-8:15 pm</b>	<b>No Class</b>

**INSTRUCTIONS for the Laboratory:**

1. Lab Score for each experiment: Your lab score will depend upon your preparedness, participation in class and your completion of the lab report. Lab reports are to be turned in at the end of class. Late report will lose points. Improper completion of lab report or improperly answered questions will lose points.
2. Print out, read, sign and return to your instructor the **Lab Safety Statement**. This statement must be signed, dated and returned by the second laboratory period (**2 Oct 2017**). The form can be found in the Course Studio under **Lab Safety Statement**.
3. You must do your laboratory work at the time assigned. Attendance will be taken. Study the experiment carefully before coming to class so that you don't (waste time going through the procedure during the lab. **NO MAKE UP LABS**).
4. You must do your own work unless you are told to work in pairs for an experiment. If you need guidance, let the instructor know.
5. Always think through the next step you are going to perform before starting it.
6. **Record all data in ink while you work**. Do not write data on paper towels or other pieces of paper, even temporarily. Make sure your data is complete. **Do not forget to write your name or record any unknown number**. Pay attention to significant figures and units. If you make a mistake, cross it out neatly with a **single line**.
7. All laboratory reports are due one week after the experiment is performed.
8. Children are not allowed in the lab.
9. No eating or drinking in the lab.
10. **ALWAYS WEAR YOUR EYE PROTECTION**. Failure to wear your eye protection will lead to dismissal from lab and a lowered grade for that experiment.
11. You **MUST WEAR LONG PANTS** and **SENSIBLE CLOTHING** when we are doing any lab that required Safety Goggles as discussed during the safety lectures. This is a school policy. If you wear shorts, sandals, or other clothing that is not consistent with safety, you will not be admitted to the laboratory. Wear a lab apron if you have one. **You can NEVER WEAR SHORT PANTS or SKIRTS during LABORATORY PERIODS.**
12. Always work with clean equipment. Clean also means **DRY**.
13. Carefully measure the quantity of each material to be used in the experiment.
14. Always place reaction vials, test tubes or flasks in a clean beaker when standing them on a laboratory bench.
15. Do not take reagent bottles to your laboratory work area. Use test tubes, beakers, or paper to obtain chemicals from the dispensing area. Take small quantities of reagents. You can always get more if you run short.
16. Check carefully the label on each reagent bottle to be sure you have the correct reagent. The names of many substances appear similar at first glance.

17. To avoid possible contamination, never return unused chemicals to the reagent bottles. Never interchange spatulas or droppers.
18. Do not insert droppers into large reagent bottles. Instead pour a little of liquid into a small beaker.
19. Be neat in your work; if you spill something, clean it up immediately.
20. Wash your hands anytime you get chemicals on them and at the end of the laboratory period.
21. Keep the mass balances and the area around them clean. Follow the directions given by the instructor on the proper weighing technique to use. Otherwise, do not place chemicals directly on the balance pans; place a piece of weighing paper or a small container on the pan first, and then weigh your material. Never weigh an object while it is hot.
22. Do not heat graduate cylinders, burettes, pipettes, or bottles with a burner flame.
23. Do not look down into the open end of a test tube in which the contents are being heated or in which a reaction is being conducted.
24. Do not perform unauthorized experiments.
25. After completing the experiment, clean and put away your glassware and equipment. Clean your work area and make sure the gas and water are turned off. A clean lab is a safe lab.
26. Dispose solid waste such as filter paper, litmus paper, and matches in the wastebasket, not in the sink. Dispose broken glass in the broken glass waste boxes. Dispose all other solid chemicals as directed by your instructor.  
**Pour all the waste liquids into the waste bottles provided or as directed by instructor. Only tap water or deionized water may be poured down the drain.**



**Student Learning Outcome(s):**

- \*Develop problem solving techniques by applying the "Scientific Method" to chemical data."
- \*Analyze and solve chemical questions utilizing information presented in the periodic table of the elements.
- \*Evaluate current scientific theories and observations utilizing a scientific mindset and an understanding of matter and the changes it undergoes.